# מכניקה קלסית וקוונטית למתמטיקאים

האוניברסיטה העברית, סתיו 1996

<u>מסי הקורס:</u> 80718. בינתיים זהו קורס שנתי, אולם אני מנסה לחלקו לשני קורסים סמסטריאליים. <u>המרצה:</u> דרור בר-נתן, בנין אינשטיין חדר 309, טלפון 658-4187, דואר אלקטרוני drorbn@math.huji.ac.il. שעות קבלה: ימי שלישי 10:30-11:30.

שעות הקורס: ימי חמישי 45:00-9:45 (אף אחד לא יבכה אם ימצא זמן יותר מוצלח). מקום הקורס עדיין לא נקבע.

<u>מטרת הקורס:</u> לעזור למתמטיקאים להבין את העקרונות הבסיסיים (והחשובים) ביותר בפיסיקה. השפה תהיה נוחה למתמטיקאים, אולם לא אנסה לשמור על דיוק מתמטי מלא - זו אינה המטרה.

נושאי הלימוד בסמסטר א: (תכנית אופטימית) חשבון ווריאציות ומשוואות Euler-Lagrange בכמה הקשרים. סימטריות וחוקי Poisson שימור. הניסוח של Hamilton, סוגרי Poisson וטיפ-טיפה על גיאומטריה סימפלקטית. ניסויי התאבכות ועקיפה. קוואנטיזציה עייי אינטגרציה על מסלולים וקוואנטיזציה קנונית. הסתברות קוונטית והפרדוקס של Bell. משפט Von-Neumann. מילים האוסצילטור ההרמוני. הצגות של (SO(3) ו-SU(2), אפקט Zeeman וספין. תבניות דיפרנציאליות ומשוואות SO(3). מילים

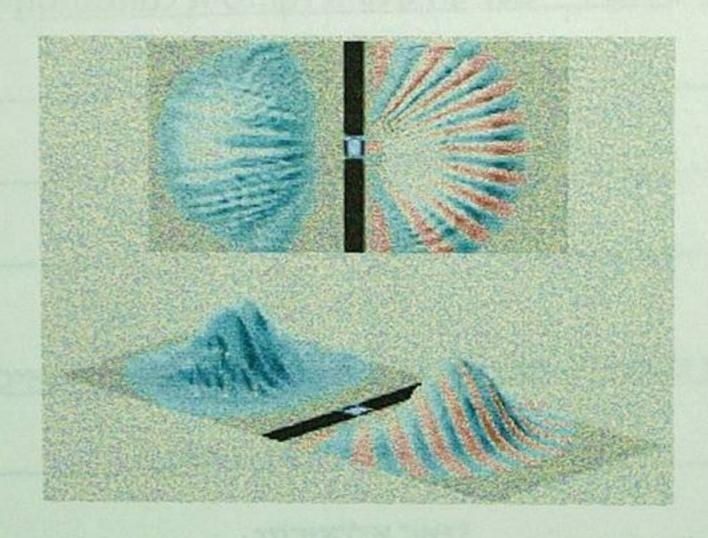
בודדות על תורת היחסות הפרטית והכללית ועל תורות שדה. <u>נושאי הלימוד בסמסטר ב:</u> אללה אכבר. נבחר איזשהו נושא קצת יותר צר ונלמד אותו יותר לעומק. מה בדיוק זה יהיה! ראו

בתחילת השורה הקודמת.

: ספרות

1. Gelfand&Fomin, Calculus of Variations.

- 2. J.J. Sakurai, Modern Quantum Mechanics.
- 3. Feynman&Hibbs, Quantum Mechanics and Path Integrals.
- 4. G. Mackey, Mathematical Foundation of Quantum Mechanics.
- 5. G. Mackey, Induced Representations of Groups and Quantum Mechanics.
- 6. Bamberg&Sternberg, A Course in Mathematics for Students of Physics.
- ספרות נוספת עייפ הצורך.



Quantum Mechanics on Quantum Theory, branch of mathematical physics that deals with the emission and absorption of energy by matter and with the motion of material particles. Because it holds that energy and matter exist in ting, discrete amounts, quantum mechanics is particularly applicable to ESEMENTARY PLARTICES and the interactions between them. According to the older theories of classical physics, energy is treated solely as a continuous phenomenon (i.e., ULAVES), and matter is assumed to occupy a very specific region of space and to move in a continuous manner. According to the quantum theory, energy is emitted and absorbed in a small packet called a quantum (pl. quanta), which in some situations behaves as particles of matter do: particles exhibit certain wavelibe properties when in motion and are no longer viewed as localized in a given region but as apread out to some degree. The quantum theory thus proposes a dual nature for both waves and particles, with one aspect predominating in some situations and the other predominating in other situations. Quantum mechanics is needed to explain many properties of matter or energy are involved, as in the interaction of elementary particles and fields, but the theory of RESLATIONISTY assumes importance in the special situation where very large speeds are involved. Together they form the theoretical basis of modern physics. (The results of classical physics appearing those of quantum mechanics for large scale events and those of relativity when ordinary speeds are involved.) Quantum theory was developed principally over a period of thirty years. The first contribution was the explanation of BSLACEKBODY radiation in 1900 by Max PSLANCEK, who proposed that the energies of any harmonic oscillator, such as the atoms of a blackbody radiator, are restricted to certain values, each of which is an integral (whole number) multiple of a basic minimum value. In 1905 Albert E.S.N.STE.S.N. proposed that the radiation itself is also quantized, and he used the new theory to explain the PSHOTOESECTRIC EFFECT. Niels BOSFR used the quantum theory in 1918 to explain both atomic structure and atomic spectra, showing the connection between the energy levels of an atom's electrons and the frequencies of light given off and absorbed by the atom. Quantum mechanics, the final mathematical formulation of the quantum theory, was developed during the 1920s. In 1924 Louis de BROGESE proposed that particles exhibit wavelike properties. This hapothesis was confirmed experimentally in 1927 by Clinton J. Davisson and Lester St. Germer, who observed DIFFRACTION of a beam of electrons. Two different formulations of quantum mechanics were presented following de Broglie's suggestion. The wave mechanics of Envir SCHRODONGER (1926) involves the use of a mathematical entity, the wave function, which is related to the probability of finding a particle at a given point in space. The matrix mechanics of Wenner SteaseNBERG (1925) makes no mention of wave functions or similar concepts but was shown to be mathematically equivalent to Schrödinger's theory. Quantum mechanics was combined with the theory of solatistic in the formulation of PLAIN DORIC (1928), which also predicted the existence of ANTIPARTICES. A particularly important discovery of the quantum theory is the uncertainty principle. enunciated by Steisenberg in 1927, which places an absolute theoretical limit on the accuracy of centain measurements; as a result the physical state of a system could be measured exactly and used to predict future states had to be abandoned. Other developments of the theory include quantum statistics, presented in one form by Einstein and S.N. Bose (Bose-Einstein statistics, which apply to BOSONS) and in another by Dirac and Enrico FERMS (Fermi-Dirac statistics, which apply to FERMOONS); quantum electronics, which deals with interactions involving quantum energy levels and resonance as in LASERS; quantum gravitation, the quantum theory of gravitational fields; and quantum field theory. In quantum field theory, interactions between particles result from the exchange of quanta: electromagnetic forces arise from the exchange of PSHOTONS, weak made as forces (see PNEAK SNTERACTION) from the exchange of 9N AND 3 PARTICIES. strong mades forces (see STRONG INTERIACTION) from the exchange of glaces, and GRIAVITIATION from the exchange of gravitors. See also QUILANTULU ESECTROPHIAMICS: QUILANTULU CHROMODYNLLUCS

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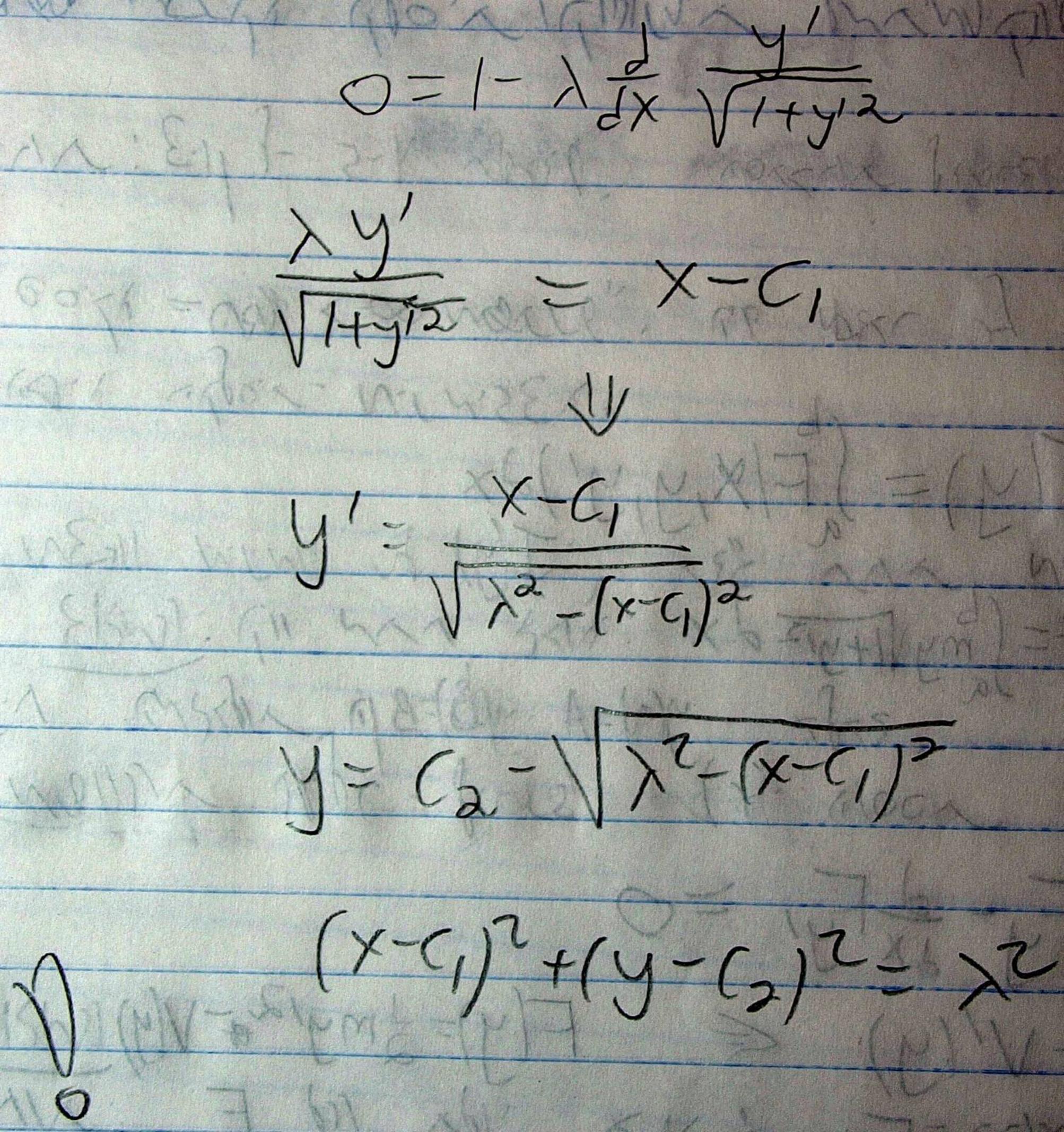
Requiring the action to be stationary leads to the generalized Euler-Lagrange equations (1-44)

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1. Use the canonical Euler equations to find the extremals of the functional

$$\int \sqrt{x^2 + y^2} \, \sqrt{1 + y'^2} \, dx,$$

and verify that they agree with those found in Chap. 1, Prob. 22.

Hint. The Hamiltonian is

$$H(x, y, p) = -\sqrt{x^2 + y^2 - p^2},$$

and the corresponding canonical system

$$\frac{dp}{dx} = \frac{y}{\sqrt{x^2 + y^2 - p^2}}, \quad \frac{dy}{dx} = \frac{p}{\sqrt{x^2 + y^2 - p^2}}$$

has the first integral

$$p^2 - y^2 = C^2$$

where C is a constant.

2. Consider the action functional

$$J[x] = \frac{1}{2} \int_{t_0}^{t_1} (mx^2 - xx^2) dt$$

corresponding to a simple harmonic oscillator, i.e., a particle of mass m acted upon by a restoring force -xx (cf. Sec. 36.2). Write the canonical system of Euler equations corresponding to J[x], and interpret them. Calculate the Poisson brackets [x, p], [x, H] and [p, H]. Is p a first integral of the canonical Euler equations?

3. Use the principle of least action to give a variational formulation of the problem of the plane motion of a particle of mass m attracted to the origin of coordinates by a force inversely proportional to the square of its distance from the origin. Write the corresponding equations of motion, the Hamiltonian and the canonical system of Euler equations. Calculate the Poisson brackets  $[r, p_r]$ ,  $[\theta, p_\theta]$ ,  $[p_r, H]$  and  $[p_\theta, H]$ , where

$$p_r = \frac{\partial L}{\partial \dot{r}}, \quad p_{\theta} = \frac{\partial L}{\partial \dot{\theta}}.$$

Is  $p_{\theta}$  a first integral of the canonical Euler equations?

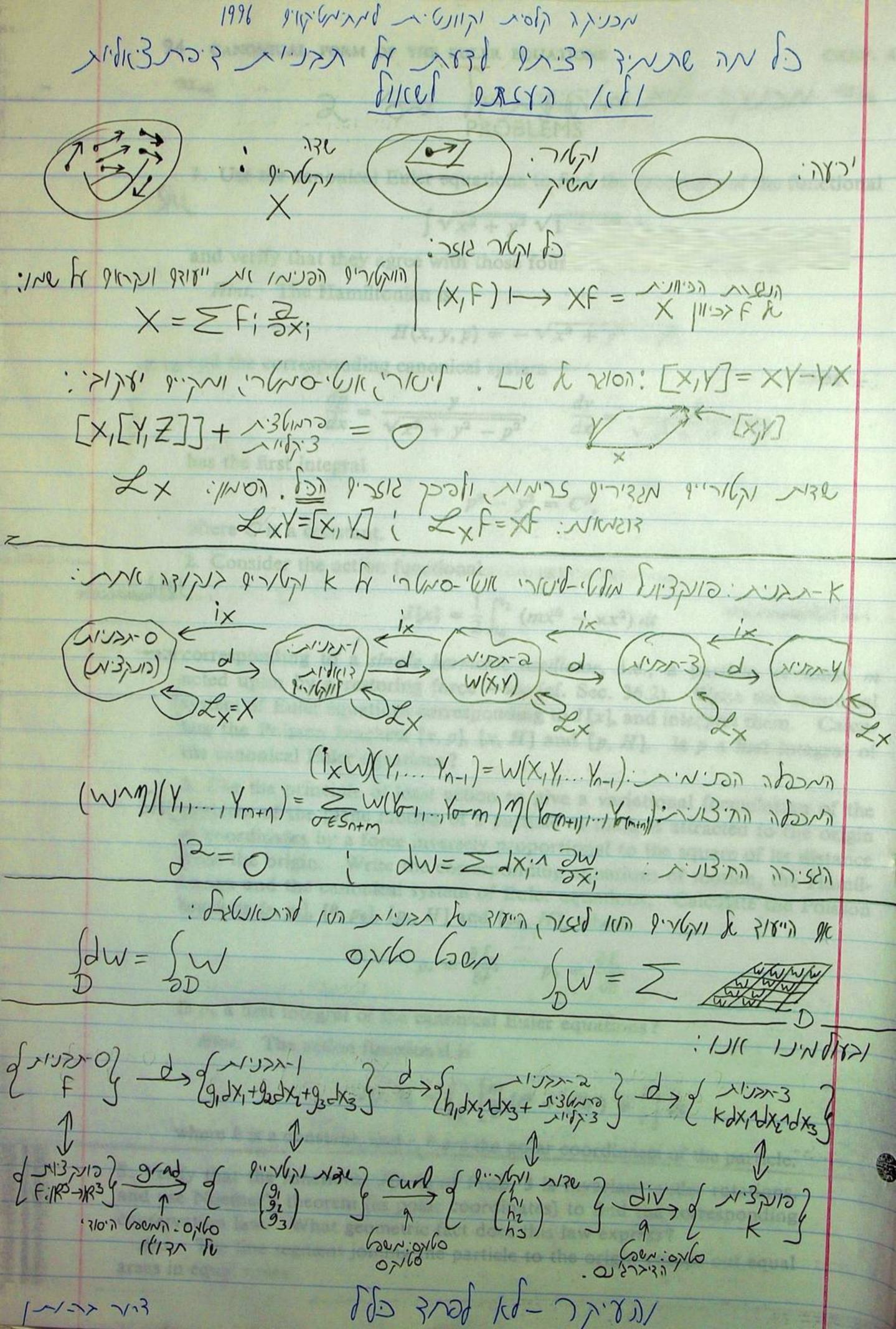
Hint. The action functional is

$$J[r,\theta] = \int_{t_0}^{t_1} \left[ \frac{m}{2} \left( \dot{r}^2 + r^2 \dot{\theta}^2 \right) + \frac{k}{r} \right] dt,$$

where k is a constant, and r,  $\theta$  are the polar coordinates of the particle.

5. Verify that the functional J[r, 0] of Prob. 3 is invariant under rotations, and use Noether's theorem (in polar coordinates) to find the corresponding conservation law. What geometric fact does this law express?

Ans. The line segment joining the particle to the origin sweeps out equal areas in equal times.



1996 JANOIS 14 (J.W.) PAS.M. (J.M.) 14 BUSHSC 9861 (2016) 2003 ACIN 8. SCHESSING.) 2. Marie onephir: M: Exx. 1911 MIIIM (d) 2010 X:25-2: W 2 dpibbii.1: NV117 ixeW=-Jf FHXF 5/1/1/1/1 [ ; p61, 43  $\left( \frac{(8y - \beta Z) d \times +}{(8y - \beta Z) d \times +} + \frac{1 \times 7 d \times}{1 + y \cdot 7 d \cdot 9} \right) d \cdot 7 + \frac{1}{1 + y \cdot 7 d \cdot 9} d \cdot 7 + \frac{1}{1$ - 95= 1 246 37455 (X JUNGS+. -(1-72) 87  $\Rightarrow \begin{pmatrix} x \\ y \end{pmatrix} = \begin{pmatrix} 4y \\ +x \\ 0 \end{pmatrix}$ {F,9}= W(xx, xy)== xx9 4351 3 H x17,7 h NUS 1000 1720 /sen.1 : Ho policy: var varif . 1 . forma +/10/10 2

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2x 14 +Qx Ax) x+2x 1/2 +Qx Az 1/2 = ( > VI-VXN)

# The Feynman Lectures on Physics, Volutt. Table 18-1 Classical Physics

### Maxwell's equations

I. 
$$\nabla \cdot E = \frac{\rho}{\epsilon_0}$$

(Flux of E through a closed surface) = (Charge inside)/ $\epsilon_0$ 

II. 
$$\nabla \times E = -\frac{\partial B}{\partial t}$$

(Line integral of E around a loop) =  $-\frac{d}{dt}$  (Flux of B through the loop)

III. 
$$\nabla \cdot \boldsymbol{B} = 0$$

(Flux of B through a closed surface) = 0

IV. 
$$c^2 \nabla \times B = \frac{j}{\epsilon_0} + \frac{\partial E}{\partial t}$$

IV.  $c^2 \nabla \times B = \frac{j}{\epsilon_0} + \frac{\partial E}{\partial t}$   $c^2$  (Integral of B around a loop) = (Current through the loop)/ $\epsilon_0$ 

$$+\frac{\partial}{\partial t}$$
 (Flux of E through the loop)

### Conservation of charge

$$\nabla \cdot j = -\frac{\partial \rho}{\partial t}$$

(Flux of current through a closed surface) =  $-\frac{\partial}{\partial t}$  (Charge inside)

Force law

$$F = q(E + v \times B)$$

Law of motion

$$\frac{d}{dt}(p) = F$$
, where

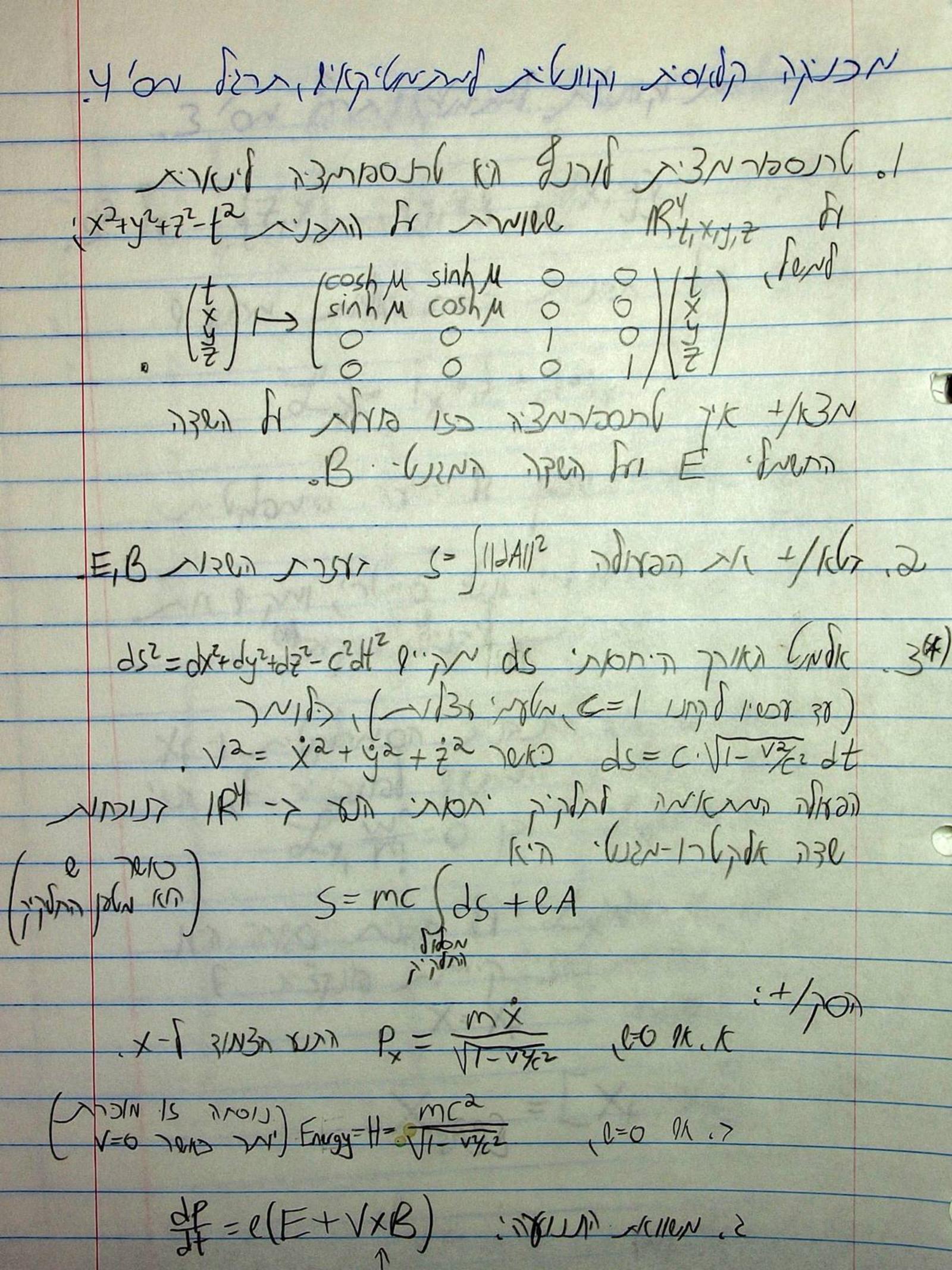
$$\frac{d}{dt}(p) = F, \quad \text{where} \quad p = \frac{mv}{\sqrt{1 - v^2/c^2}}$$

(Newton's law, with Einstein's modification)

Gravitation

$$F = -G \frac{m_1 m_2}{r^2} e_r$$

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# Non-Commutative (Quantum) Probability Classical and Quantum Mechanics for Mathematicians, HUJI 1996 Dror Bar-Natan

Claim: In the quantum probability space  $(\mathbb{R}^4, v)$  where v is the unit vector  $v = \frac{\sqrt{2}}{2}(0 \ 1 \ -1 \ 0)^T$ , one has  $p(A = B) = p(B = C) = p(C = D) = \frac{3}{4}$  and p(D = A) = 0, where A, B, C, and D are the random variables corresponding to the matrices:

$$A = \begin{pmatrix} 1 & 0 & 0 & 0 \\ 0 & 1 & 0 & 0 \\ 0 & 0 & -1 & 0 \\ 0 & 0 & 0 & -1 \end{pmatrix} \quad ; \quad B = \begin{pmatrix} -\frac{1}{2} & \frac{\sqrt{3}}{2} & 0 & 0 \\ \frac{\sqrt{3}}{2} & \frac{1}{2} & 0 & 0 \\ 0 & 0 & -\frac{1}{2} & \frac{\sqrt{3}}{2} \\ 0 & 0 & \frac{\sqrt{3}}{2} & \frac{1}{2} \end{pmatrix}$$

$$C = \begin{pmatrix} -\frac{1}{2} & 0 & -\frac{\sqrt{3}}{2} & 0 \\ 0 & -\frac{1}{2} & 0 & -\frac{\sqrt{3}}{2} \\ -\frac{\sqrt{3}}{2} & 0 & \frac{1}{2} & 0 \\ 0 & -\frac{\sqrt{3}}{2} & 0 & \frac{1}{2} \end{pmatrix} \quad ; \quad D = \begin{pmatrix} 1 & 0 & 0 & 0 \\ 0 & -1 & 0 & 0 \\ 0 & 0 & 1 & 0 \\ 0 & 0 & 0 & -1 \end{pmatrix}.$$

Mathematica 2.0 for SPARC Copyright 1988-91 Wolfram Research, Inc. -- Terminal graphics initialized --

In[1]:= v=1/2 Sqrt[2] {0, 1, -1, 0}; q=1/2 Sqrt[3];

In[2]:= A1=DiagonalMatrix[{1, 1, -1, -1}]; A4=DiagonalMatrix[{1, -1, 1, -1}];

In[3]:=  $A2=\{\{-1/2, q, 0, 0\}, \{q, 1/2, 0, 0\}, \{0, 0, -1/2, q\}, \{0, 0, q, 1/2\}\};$ 

In[4]:= A3={ $\{-1/2, \bar{0}, -q, 0\}, \{\bar{0}, -1/2, 0, -q\}, \{-q, 0, 1/2, \bar{0}\}, \{0, -q, \bar{0}, 1/2\}\};$ 

In[5]:= {Eigenvalues[A1], Eigenvalues[A2], Eigenvalues[A3], Eigenvalues[A4]}

Out  $[5] = \{\{1, -1, 1, -1\}, \{1, -1, 1, -1\}, \{1, -1, 1, -1\}, \{1, -1, 1, -1\}\}$ 

 $In[6] := \{A1.A2 == A2.A1, A2.A3 == A3.A2, A3.A4 == A4.A3, A4.A1 == A1.A4\}$ 

Out[6] = {True, True, True, True}

 $In[7] := pequal[M1_, M2_] := 1-v.(M1-M2).(M1-M2).v / 4$ 

In[8]:= {pequal[A1,A2], pequal[A2,A3], pequal[A3,A4], pequal[A4,A1]}

3 3 3 Out[8]= {-, -, -, 0} 4 4 4

More information can be found at N.D. Mermin, Physics Today 39(4) 38 (1985) and D. Bar-Natan, Foundations of Physics 19(1) 97 (1989).

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### Quantum Behavior

#### 1-1 Atomic mechanics

"Quantum mechanics" is the description of the behavior of matter and light in all its details and, in particular, of the happenings on an atomic scale. Things on a very small scale behave like nothing that you have any direct experience about. They do not behave like waves, they do not behave like particles, they do not behave like clouds, or billiard balls, or weights on springs, or like anything that you have ever seen.

Newton thought that light was made up of particles, but then it was discovered that it behaves like a wave. Later, however (in the beginning of the twentieth century), it was found that light did indeed sometimes behave like a particle. Historically, the electron, for example, was thought to behave like a particle, and then it was found that in many respects it behaved like a wave. So it really behaves like neither. Now we have given up. We say: "It is like neither."

There is one lucky break, however—electrons behave just like light. The quantum behavior of atomic objects (electrons, protons, neutrons, photons, and so on) is the same for all, they are all "particle waves," or whatever you want to call them. So what we learn about the properties of electrons (which we shall use for our examples) will apply also to all "particles," including photons of light.

The gradual accumulation of information about atomic and small-scale behavior during the first quarter of this century, which gave some indications about how small things do behave, produced an increasing confusion which was finally resolved in 1926 and 1927 by Schrödinger, Heisenberg, and Born. They finally obtained a consistent description of the behavior of matter on a small scale. We take up the main features of that description in this chapter.

Because atomic behavior is so unlike ordinary experience, it is very difficult to get used to, and it appears peculiar and mysterious to everyone—both to the novice and to the experienced physicist. Even the experts do not understand it the way they would like to, and it is perfectly reasonable that they should not, because all of direct, human experience and of human intuition applies to large objects. We know how large objects will act, but things on a small scale just do not act that way. So we have to learn about them in a sort of abstract or imaginative fashion and not by connection with our direct experience.

In this chapter we shall tackle immediately the basic element of the mysterious behavior in its most strange form. We choose to examine a phenomenon which is impossible, absolutely impossible, to explain in any classical way, and which has in it the heart of quantum mechanics. In reality, it contains the only mystery. We cannot make the mystery go away by "explaining" how it works. We will just tell you how it works. In telling you how it works we will have told you about the basic peculiarities of all quantum mechanics.

### 1-2 An experiment with bullets

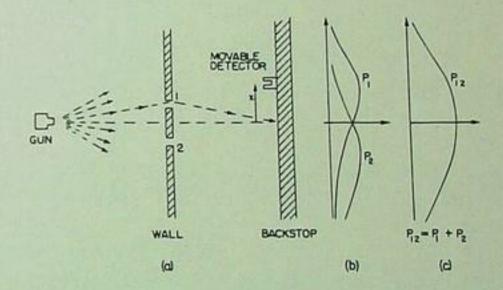
To try to understand the quantum behavior of electrons, we shall compare and contrast their behavior, in a particular experimental setup, with the more familiar behavior of particles like bullets, and with the behavior of waves like water waves. We consider first the behavior of bullets in the experimental setup shown diagrammatically in Fig. 1-1. We have a machine gun that shoots a stream of bullets. It is not a very good gun, in that it sprays the bullets (randomly) over a fairly large angular spread, as indicated in the figure. In front of the gun we have

- 1-1 Atomic mechanics
- 1-2 An experiment with bullets
- 1-3 An experiment with waves
- 1-4 An experiment with electrons
- 1-5 The interference of electron waves
- 1-6 Watching the electrons
- 1-7 First principles of quantum mechanics
- 1-8 The uncertainty principle

Note: This chapter is almost exactly the same as Chapter 37 of Volume I.

Fig. 1-1. Interference experiment with bullets.

2 wall (made of armor plate) that has in it two holes just about big enough to let a bullet through. Beyond the wall is a backstop (say a thick wall of wood) which will "absorb" the bullets when they hit it. In front of the wall we have an object which we shall call a "detector" of bullets. It might be a box containing sand. Any bullet that enters the detector will be stopped and accumulated. When we wish, we can empty the box and count the number of bullets that have been caught. The detector can be moved back and forth (in what we will call the x-direction). With this apparatus, we can find out experimentally the answer to the question: "What is the probability that a bullet which passes through the holes in the wall will arrive at the backstop at the distance x from the center?" First, you should realize that we should talk about probability, because we cannot say definitely where any particular bullet will go. A bullet which happens to hit one of the holes may bounce off the edges of the hole, and may end up anywhere at all. By "probability" we mean the chance that the bullet will arrive at the detector, which we can measure by counting the number which arrive at the detector in a certain time and then taking the ratio of this number to the total number that hit the backstop during that time. Or, if we assume that the gun always shoots at the same rate during the measurements, the probability we want is just proportional to the number that reach the detector in some standard time interval.



For our present purposes we would like to imagine a somewhat idealized experiment in which the bullets are not real bullets, but are indestructible bulletsthey cannot break in half. In our experiment we find that bullets always arrive in lumps, and when we find something in the detector, it is always one whole bullet. If the rate at which the machine gun fires is made very low, we find that at any given moment either nothing arrives, or one and only one-exactly one-bullet arrives at the backstop. Also, the size of the lump certainly does not depend on the rate of firing of the gun. We shall say: "Bullets always arrive in identical lumps." What we measure with our detector is the probability of arrival of a lump. And we measure the probability as a function of x. The result of such measurements with this apparatus (we have not yet done the experiment, so we are really imagining the result) are plotted in the graph drawn in part (c) of Fig. 1-1. In the graph we plot the probability to the right and x vertically, so that the x-scale fits the diagram of the apparatus. We call the probability  $P_{12}$  because the bullets may have come either through hole 1 or through hole 2. You will not be surprised that  $P_{12}$  is large pear the middle of the graph but gets small if x is very large. You may wonder, however, why  $P_{12}$  has its maximum value at x = 0. We can understand this fact if we do our experiment again after covering up hole 2, and once more while covering up hole 1. When hole 2 is covered, bullets can pass only through hole 1, and we get the curve marked P1 in part (b) of the figure. As you would expect, the maximum of  $P_1$  occurs at the value of x which is on a straight line with the gun and hole 1. When hole 1 is closed, we get the symmetric curve P2 drawn in the figure. P2 is the probability distribution for bullets that pass through hole 2. Comparing parts (b) and (c) of Fig. 1-1, we find the important result that

2007 1910 V 161117, June 19 18 ENCL 966/ 1098. Reed-simon chap ITIIII , ryien must its per in super in since inside with  $(e^{-9}(p)(q)-4(q+9.))$  ?  $e^{-9}(p)$  in P=-iq wel? 10 dr 6 dy -1 east 14(1, 19 dr dy -1 dx 100 R (Fig. 1.1. The thing of the stand of the stand of the stand of the stands of the stan (Wyl In hom on) Leve = e'xp Vp Ly Coen JK . ~ 37/3733 JUN DIS IN JURD = Q+XI , N'UM MYA UVBU- = exp(V\_x(ipQ)U-x) = . TO NO NS: 210" NMM S FI Q=X-id P=-id x 220 MAR (21, Y, A(t), B(t), ) = = -i[A,H]: ">>>1) = = -i[A,H]: ">>>1) = >10 (-xflipo 25 for if 132W e-ith). A(t) = e+ith a. eith Strike U pr ( AP, UV, UAU, U'BV,...) ( AP, V, A,B,...) 1000 1 4 0 0 = e-ith. 4. 1) : JUNG: WAY, A,B...) : JUNG: 27 JUNG: 27000 7) 17 NULL = + iH4 1/2/2 1.75/2 以=VInix X=病(年录)かとが2 1つか、t=至一、共=一(音器2 + 2 年) 1 120 (4) JYn=F(A-8/9) Yn==-1(A-1) FYn-1

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כל המשתנים מוצנים על בסיס שנתי. סטיית תקן גלומה – סטיית התקן אשר בהצבתה בנוסחת B&S נקבל את מחיר הסנירה. DELTA – שעור השינוי בשווי האופציה העבע מעליית מדד המערף בנקודה – בהנחה ששאר המשתנים הקובעים את שווי האופציה נשארים קבועים. THETA – שעור השינוי בשווי האופציה העבע מקיצור התקופה למימוש ביום אחד – בהנחה ששאר המשתנים הקובעים את שווי האופציה נשארים קבועים. – עושר השינוי בשווי האופציה הנובע מעליית סמיית התקן באחוז – בהנחה ששאר המשתנים הקובעים את שווי האופציה נשארים קבועים.

# מכניקה קלסית וקוונטית למתמטיקאים

האוניברסיטה העברית, אביב 1997

<u>מס' הקורס:</u> 80719.

<u>המרצה:</u> דרור בר-נתן, בנין אינשטיין חדר 309, טלפון 202-658-4187, דואר אלקטרוני drorbn@math.huji.ac.il <u>המרצה:</u> כל יום, כל היום, אבל רק אם אני פנוי.

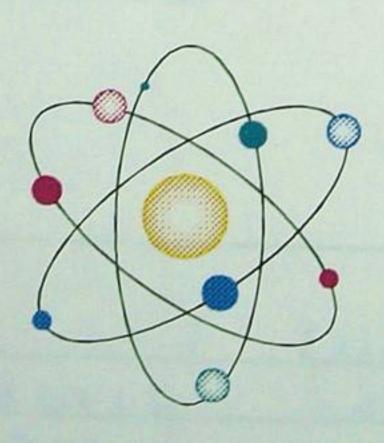
<u>שעות הקורס:</u> ימי רביעי 14:00-16:00 בינתיים. ננסה לשפר ע"פ דרישות הנוכחים ואולי גם לפצל לשתי פגישות שבועיות.

מטרת הקורס: לעזור למתמטיקאים להבין את העקרונות הבסיסיים (והחשובים) ביותר בפיסיקה. השפה תהיה נוחה למתמטיקאים, אולם לא אנסה לשמור על דיוק מתמטי מלא - זו אינה המטרה. הכלי: ננסה להבין עד כמה שניתן את הספקטרום של אטום המימן. אני מקווה שהנושא ישמור עלינו מרוכזים בכיוון אחד, אבל גם יחייב אותנו ללמוד ולהבין את הפיסיקה הרבה שמסביבו.

<u>מטרה משנית:</u> המרצה עצמו רוצה ללמוד את כל הנושאים האלו. חזית הידע שלו כרגע היא מקבילה לחומר הנדרש. אילו היא הייתה ניצבת, היינו במצב חסר סיכוי. אבל כך, כל שנדרש מהמרצה הוא מאמץ נוסף לאורך מספר רב של תחומים מפוזרים היטב בזמן, ויש תקוה.

<u>השיטה:</u> צירוף המטרה העיקרית והמטרה המשנית יחייב הרבה שיתוף פעולה מצד הסטודנטים! ננסה לעקוב ביחד אחרי Everything You Always Wanted to Know About the Hydrogen Atom (But Were Afraid to Ask) המאמר (ראי של אתר האינטרנט של המאמר נמצא ב- http://tipesh.ma.huji.ac.il) וללמוד לאורך הדרך את כל מה שנצטרך ללמוד.

נושאי הלימוד: (תכנית אופטימית) השלמת אינטגרציה על מסלולים מסמסטר א', חשבונות קוואנטיים ישנים ע"פ Bohr, מרכז המסה, קרינה אלקטרומגנטית, פתרון משוואת Schrodinger והקשר עם הצגות של (SO(3), חשבונות ספין ארכז המסה, קרינה אלקטרומגנטית, פתרון משוואת Klein-Gordon וחשבונות יחסותיים נאיוויים, משוואת של הבילות ולשרים (אם כבר אנחנו כאן) נירמול מחדש, אלקטרודינמיקה קוואנטית, Chern-Simons ושמורות של יריעות וקשרים (אם כבר אנחנו כאן) נירמול מחדש, אלקטרודינמיקה קוואנטית, ההזזה של Lamb, אפקטים נוספים.



Hydrogen, (H), gaseous element, discovered by Henry CAVENDISH in 1766. The first element on the PERIODIC TABLE, hydrogen is colorless, odorless, tasteless, slightly soluble in water, and highly explosive. The hot flame produced by a mixture of oxygen and hydrogen is used in welding and in melting quartz and glass. Normal hydrogen is diatomic (see protium (nucleus: one proton), the most abundant element in the universe, hydrogen is the major fuel in fusion reactions of the SUN and other STARS. Atmospheric hydrogen has three isotopes: chemical-reaction mechanisms; and tritium (nucleus: one proton and two neutrons), a radioactive gas used in the hydrogen bomb, in luminous paints, and as a tracer. Hydrogen has three restrictions are the symbolic of ANACONIA. It is the symbolic of the symboli

# Everything You Always Wanted to Know About the Hydrogen Atom (But Were Afraid to Ask)

Randal C. Telfer Johns Hopkins University

May 6, 1996

### Abstract

A thorough review of the structure of the hydrogen atom will be presented with emphasis on the quantum-mechanical principles involved rather than calculational detail, which will be minimized. First, the relationship of the Heisenberg uncertainty principle to the hydrogen atom will be discussed briefly. This is followed by a discussion of the energy level structure of the hydrogen atom, including fine structure, in the context of the quantum-mechanical theories of Bohr, Schrödinger, and Dirac. Finally, smaller-order corrections to these theories will be discussed, including the Lamb shift, hyperfine structure, and the Zeeman effect.

## 1 The Uncertainty Principle

Before discussing specifics about the structure of the hydrogen atom, it is interesting to note what information about the hydrogen atom can be derived just from the Heisenberg uncertainty principle. A familiar form of the uncertainty principle looks like the following:

$$\Delta x \Delta p_x \sim \hbar,$$
 (1)

where  $\Delta x$  and  $\Delta p_x$  are the uncertainty in the x-component of the position and momentum of a particle, respectively. Consider an electron in a classical circular orbit in the xy-plane. It is then reasonable to write  $\Delta x \sim r$ , where r is the radius of the orbit. Assuming a state of minimum uncertainty,  $\Delta p_x$  is then known from the uncertainty principle, and it should be roughly equal

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QUANTITY	NAME OF BASE SL UNIT	SYMBO
Length	meter (or metre)	m
Mass	kilogram	kg
Time	second	5
Electric current	ampere	Α
Thermodynamic temperature	kelvin	K
Amount of substance	mole	mol
Luminous intensity	candela	cd
TABLE 2	NAME OF SUPPLEMENTARY SL UNIT	SYMBO
TABLE 2	SUPPLEMENTARY	SYMB0

Tables 1 and 2: International System of Units (SI Units)

International System of Units (French Le SystUme International d'Unit's), name adopted by the Eleventh General Conference on Weights and Measures, held in Paris in 1960, for a universal, unified, self-consistent system of measurement units based on the MKS (meter-kilogram-second) system. The international system is commonly referred to throughout the world as SI, after the initials of SystUme International. The Metric Conversion Act of 1975 commits the U.S. to the increasing use of, and voluntary conversion to, the metric system of measurement, further defining metric system as the International System of Units as interpreted or modified for the U.S. by the secretary of commerce.

At the 1960 conference, standards were defined for six base units and for two supplementary units; a seventh base unit, the mole, was added in 1971. The seven base units are listed in Table 1, and the supplementary units are listed in Table 2. The symbols in the last column are not abbreviations (hence, no periods are used), and they are exactly the same in all languages.

#### Length

The meter and the kilogram had their origin in the metric system. By international agreement, the standard meter had been defined as the distance between two fine lines on a bar of platinum-iridium alloy. The 1960 conference redefined the meter as 1,650,763.73 wavelengths of the reddish-orange light emitted by the isotope krypton-86. The meter was again redefined in 1983 as the length of the path traveled by light in vacuum during a time interval of 1/299,792,458 of a second.

### Mass

When the metric system was created, the kilogram was defined as the mass of 1 cubic decimeter of pure water at the temperature of its maximum density (4.0° C/39.2° F). A solid cylinder of platinum was carefully made to match this quantity of water under the specified conditions. Later it was discovered that a quantity of water as pure or as stable as required could not be provided. Therefore the primary standard of mass became the platinum cylinder, which was replaced in 1889 by a platinum-iridium cylinder of similar mass. Today this cylinder still serves as the international kilogram, and the kilogram in SI is defined as a quantity of mass of the international prototype of the kilogram.

#### Time

For centuries, time has been universally measured in terms of the rotation of the earth. The second, the basic unit of time, was defined as 1/86,400 of a mean solar day (see DAY) or one complete rotation of the earth on its

axis. Scientists discovery owever, that the rotation of the earth was not constant enough to serve as the basis of the time standard. As a result is second was redefined in 1967 in terms of the resonant frequency of the cesium atom, that is, the frequency at which this atom absorbs energy: 9,192,631,770 Hz (cycles per second).

### Temperature

The temperature scale adopted by the 1960 conference was based on a fixed temperature point, the triple point of water, at which the solid, liquid, and gas are in equilibrium. The temperature of 273.16 K was assigned to this point. The freezing point of water was designated as 273.15 K, equaling exactly 0° on the Celsius temperature scale. The Celsius scale, which is identical to the centigrade scale, is named for the 18th-century Swedish astronomer Anders Celsius, who first proposed the use of a scale in which the interval between the freezing and boiling points of water is divided into 100 degrees. By international agreement, the term Celsius has officially replaced centigrade.

### Other Units

TABLE 3			VALUE IN TERMS O UNITS OR IN TERM			
	Where a serious		SPECIAL NAME	OF DERIVED	SLUNIT	
DUANTITY	NAME OF DERIVED SI UNIT	SYMBOL	QUANTITY	S	YMBOL	
Area	square meter	m²	Frequency	hertz	Hz	1/5
Volume	cubic meter	m³	Force	newton	N	kg-m/s2
Velocity	meter per second	m/s	Pressure, stress	pascal	Pa	N/m²
Acceleration	meter per second squared	m/s2	Energy, work, quantity of heat	joule	J	N-m
Density	kilogram per cubic meter	kg/m <sup>3</sup>	Power	watt	W	J/s
Current density	ampere per square meter	A/m <sup>2</sup>	Quantity of electricity	coulomb	C	A-s
Magnetic field strength	ampere per meter	A/m	Electric potential	volt	٧	W/A
Specific volume	cubic meter per kilogram	m <sup>3</sup> /kg	Capacitance	farad	F	G/V
Luminance	candela per square meter	cd/m <sup>2</sup>	Electric resistance	ohm	Ω	V/A
		100	Conductance	siemens	S	A/V
			Magnetic flux	weber	Wb	V.5
			Magnetic flux density	tesla	Т	Mp/ws
			Inductance	henry	н	Wb/A
			Luminous flux	lumen	1m	cd-sr
			Muminance	lux	1×	Jm/ms
			Activity (of radionuclides)	becquere'	l Bq	1/5
icrosoft Table			Absorbed dose	gray	Gy	J/kg

Tables 3 and 4: Derived SI Units and Names

MULTIPLICATION FACTOR		PREFIX	SYMBOL
1000 000 000 000 000 000	= 1018	exa	E
1000 000 000 000 000	= 1015	peta	P
1000 000 000 000	= 1012	tera	T
1000 000 000	= 109	giga	G
1000 000	= 106	mega	M
1000	1 = 103	kilo	k
100	102	hecto	h
10	101	deka	da
0.1	= 10-1	deci	d
0.01	= 10-2	centi	0
0.001	= 10-3	milli	m
0.000 001	1 = 10-6	micro	*
0.000 000 001	1 = 10-9	nano	n
0.000 000 000 001	1 = 10-12	pico	P
0.000 000 000 000 00	1 = 10-15	femto	f
0 000 000 000 000 000 00	1 = 10-18	atto	

Table 5: Metric Prefixes

QUANTITY	NAME OF UNIT	UNIT SYMBOL	DEFINITION
Time	minute	min	1 min = 60 s
	hour	h -	1 h = 60 min
	day	d	1 day = 24 h
Plane	degree	•	$1^{\circ} = (\pi/180)$
angle			rad
aligic	minute	1 3	1' = (1/60)°
	second	11	1" = (1/60)"
Volume	liter (or litre)	1	1 l = 1 dm <sup>3</sup>
Mass	metric to	n t	1 t = 10³kg

Table 6: Standard Quantities and Units

In SI, the ampere was defined as the constant current that, flowing in two parallel conductors one meter apart in a vacuum, will produce a force between the conductors of 2 × 10-7 newtons per meter of length.

In 1971 the mole was defined as the amount of substance of a system that contains as many elementary entities as there are atoms in 0.012 kilogram of carbon-12.

The international unit of light intensity, the candela, was defined as 1/60 of the light radiated from a square centimeter of a blackbody, a perfect radiator that absorbs no light, held at the temperature of freezing platinum. The radian is the plane angle between two radii of a circle that cut off on the circumference an arc equal in length to the radius.

The steradian is defined as the solid angle that, having its vertex in the center of a sphere, cuts off an area of the surface of the sphere equal to that of a square with sides of length equal to the radius of the sphere.

The SI units for all other quantities are derived from the seven base units and the two supplementary units. Examples of some SI-derived units, expressed in terms of base units, are shown in Table 3. Some derived units are used so often that they have been assigned special names—usually those of scientists—as shown in Table 4. One feature of SI is that it is a coherent system; that is, derived units are expressed as products and ratios of the base, supplementary, and other derived units without numerical factors. This results in some units being too large for ordinary use and others too small. To compensate, the prefixes developed for the metric system have been borrowed and expanded. These prefixes, given in Table 5, are used with all three types of units: base, supplementary, and derived. Examples are millimeter (mm), kilometer/hour (km/h), megawatt (MW), and picofarad (pF). Because double prefixes are not used, and because the base unit kilogram already contains a prefix, prefixes are not used with kilogram, although they are used with gram. The prefixes hecto, deka, deci, and centi are used only rarely, and then usually with meter to express areas and volumes. Because of established usage, the centimeter is retained for body measurements and clothing.

Certain units that are not part of SI are used so widely that it is impractical to abandon them. The units that are accepted for continued use in the U.S. with SI are listed in Table 6.

In cases where their usage is already well established, certain other units are allowed for a limited time, subject to future review: nautical mile, knot, angstrom, standard atmosphere, hectare, and bar. See also METRIC SYSTEM; WEIGHTS AND MEASURES.

### Further Reading

"International System of Units," Microsoft (R) Encarta. Copyright (c) 1994 Microsoft Corporation. Copyright (c) 1994 Funk & Wagnall's Corporation.

Electrical Units, units used to express quantitative measurements of all types of electrostatic and electromagnetic phenomena and of the electrical characteristics of components of electrical circuits . The basic electrical units are part of the centimeter-gram-second system, but because, in most cases, these units are either too large or too small for convenient measurement, a number of practical units have been adopted for use in engineering.

#### Basic Units

The elemental unit of electricity is the absolute charge on a single electron or proton. The symbol for this unit is e. The CGS unit of electrical charge is the electrostatic unit (esu), which is defined as the quantity of electricity that when concentrated at a point in a vacuum will repel a like charge 1 cm away with a force of 1 dyne. The esu equals the aggregate charge carried by 2,082,000,000 electrons or protons.

The basic units of electrical current or flow is the statampere, which is defined as a current of 1 esu/sec. The statvolt, the basic unit of electromotive force, or potential difference, is the difference in potential that exists between two points when 1 erg of work is required to force 1 esu of electricity between those two points.

### Electromagnetic Units

Besides the electrostatic units of charge, current, and potential difference, a parallel group of basic electromagnetic units exists. The basic magnetic unit, comparable to the elemental unit of electricity, is the unit magnetic pole, defined as a point magnetic pole that in a vacuum will act on a similar pole 1 cm away with a force of 1 dyne. The unit used to measure the strength of magnetic fields is the oersted. A field that acts on a unit magnetic pole with a force of 1 dyne has a strength of 1 oersted. The electromagnetic unit of electric current is called the abampere. If a current of 1 abampere flows in a wire 1 cm long, the wire is pushed sidewise with a force of 1 dyne by a magnetic field of 1 oersted acting at right angles to the wire. The abcoulomb is the quantity of electricity passing any point in a circuit in 1 sec when a current of 1 abampere is flowing in the circuit. The abvolt, the electromagnetic unit of potential difference, is the potential difference between two points when 1 erg of work is necessary to move 1 abcoulomb of electricity from one point to the other. See also POTENTIAL ENERGY.

The mathematical relationships between the electrostatic and electromagnetic units are as follows: I esu equals 3.3356 × 10-11 abcoulombs; 1 statampere equals 3.3356 × 10-11 abamperes; and one statvolt equals 29,979,600,000 abvolts. This last figure is exactly equal to the velocity of light through a vacuum, which is expressed in centimeters per second, as predicted by the electromagnetic-wave theory developed by the British physicist James Clerk Maxwell. See ELECTROMAGNETIC RADIATION.

#### Practical Units

The unit of electrical current in common use is the ampere, which is defined as 0.1 abamperes. The practical unit of electrical quantity is the coulomb, the amount of electricity passing a given point in a circuit in 1 sec when a current of 1 amp is flowing. The volt is the practical unit of potential difference. It is equal to 100 million abvolts and can be defined as the potential difference existing between two points when 1 joule (10 million ergs) of work is required to move I coulomb of electricity from one of the points to the other. The unit of electrical work is the watt. It represents the generation or use of electrical energy at the rate of 1 J/sec. The kilowatt is equal to 1000 watts.

Because of the difficulty of making measurements in terms of the absolute units, the practical units are also defined for purposes of practical standardization as follows: The ampere is the amount of current that will deposit 0.001118 g of silver per sec if passed through a silver nitrate solution; the ohm is the resistance of a column of mercury 106.3 cm in length and 1 sq mm in cross section at a temperature of 0° C (32° F); the volt is the electromotive force necessary to produce a current of 1 amp through a resistance of 1 ohm. The volt is also defined in terms of a standard voltaic cell, called the Weston cell, which has poles of cadmium amalgam and mercurous sulfate and an electrolyte of cadmium sulfate. A volt is defined as 0.98203 of the potential of this standard cell at 20° C (68° F).

In all the practical electrical units the conventional prefixes of the metric system are used to indicate fractions and multiples of the basic units. Thus a micromicrofarad is a trillionth of a farad, a microampere is a millionth of an ampere, a millivolt is a thousandth of a volt, a millihenry is a thousandth of a henry, a kilowatt is 1000 watts, and a megohm is 1 million ohms.

### Resistance, Capacitance, Inductance

All components in electrical circuits exhibit one or more of the characteristics of resistance, capacitance, and inductance. The commonly used unit of resistance is the ohm, which is the resistance of a conductor in which a potential difference of 1 V causes a current flow of 1 amp. The capacitance of a condenser is measured in farads. A condenser of 1 farad capacitance will exhibit a change in potential difference of 1 V between its plates when 1 coulomb of electricity is transferred from one plate to the other. The henry is the unit of inductance. A coil has a self-inductance of 1 H when a change in current of 1 amp/sec produces a countervoltage of 1/V. In a transformer, or in any two magnetically coupled circuits, a mutual induction of 1 H is that inductance which will induce a voltage of 1 V in the secondary when there is a change of 1 amp/sec in the primary.

See also BATTERY; ELECTROCHEMISTRY.

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Planck's Constant, fundamental physical constant, symbol h. It was first discovered (1900) by the German physicist Max Planck. Until that year, light in all forms had been thought to consist of waves. Planck noticed certain deviations from the wave theory of light on the part of radiations emitted by so-called blackbodies, or perfect absorbers and emitters of radiation. He came to the conclusion that these radiations were emitted in discrete units of energy, called quanta. This conclusion was the first enunciation of the quantum theory. According to Planck, the energy of a quantum of light is equal to the frequency of the light multiplied by a constant. His original theory has since had abundant experimental verification, and the growth of the quantum theory has brought about a fundamental change in the physicist's concept of light and matter, both of which are now thought to combine the properties of waves and particles. Thus, Planck's constant has become as important to the investigation of particles of matter as to quanta of light, now called photons. The first successful measurement (1916) of Planck's constant was made by the American physicist Robert Millikan. The present accepted value of the constant is h = 6.626 × 10-34 joule-second in the meter-kilogram-second system.

"Planck's Constant," Microsoft (R) Encarta. Copyright (c) 1994 Microsoft Corporation. Copyright (c) 1994 Funk & Wagnall's Corporation.

Electrons play a role in a wide variety of phenomena. The flow of an electric current in a conductor is caused by the drifting of free electrons in the conductor. Heat conduction is also primarily a phenomenon of electron activity. In vacuum tubes a heated cathode emits a stream of electrons that can be used to amplify or rectify an electric current (see RECTIFICATION; VACUUM TUBES). If such a stream is focused into a well-defined beam, it is called a cathode-ray beam (see CATHODE RAY). Cathode rays directed against suitable targets produce X rays; directed against the fluorescent screen of a television tube, they produce visible images. Also, the negatively charged beta particles emitted by some radioactive substances are electrons. See RADIOACTIVITY; X RAY; ELECTRONICS; PARTICLE ACCELERATORS.

Electrons have a rest mass of 9.109 x 10-28 grams, and an electrical charge of negative 1.602 x 10-19 coulombs (see ELECTRICAL UNITS). The charge of the electron is the basic unit of electricity. Electrons are classified as fermions because they have half-integral spin; spin is a quantum mechanical property of subatomic particles that indicates the particle's angular momentum. The antimatter version of the electron is the positron.

Contributed by: Richard Hofstadter

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In[1]:= << Miscellaneous PhysicalConstants</pre>

In[2]:= ElectronCharge

Out[2] = 1.60218 10 Coulomb

In[3]:= ElectronMass

-31Out[3] = 9.10939 10 Kilogram

In[4]:= PlanckConstantReduced

Out[4] = 1.05457 10 Joule Second

In[5]:= VacuumPermittivity

8.85419 Ampere Second
Out[5] = ------

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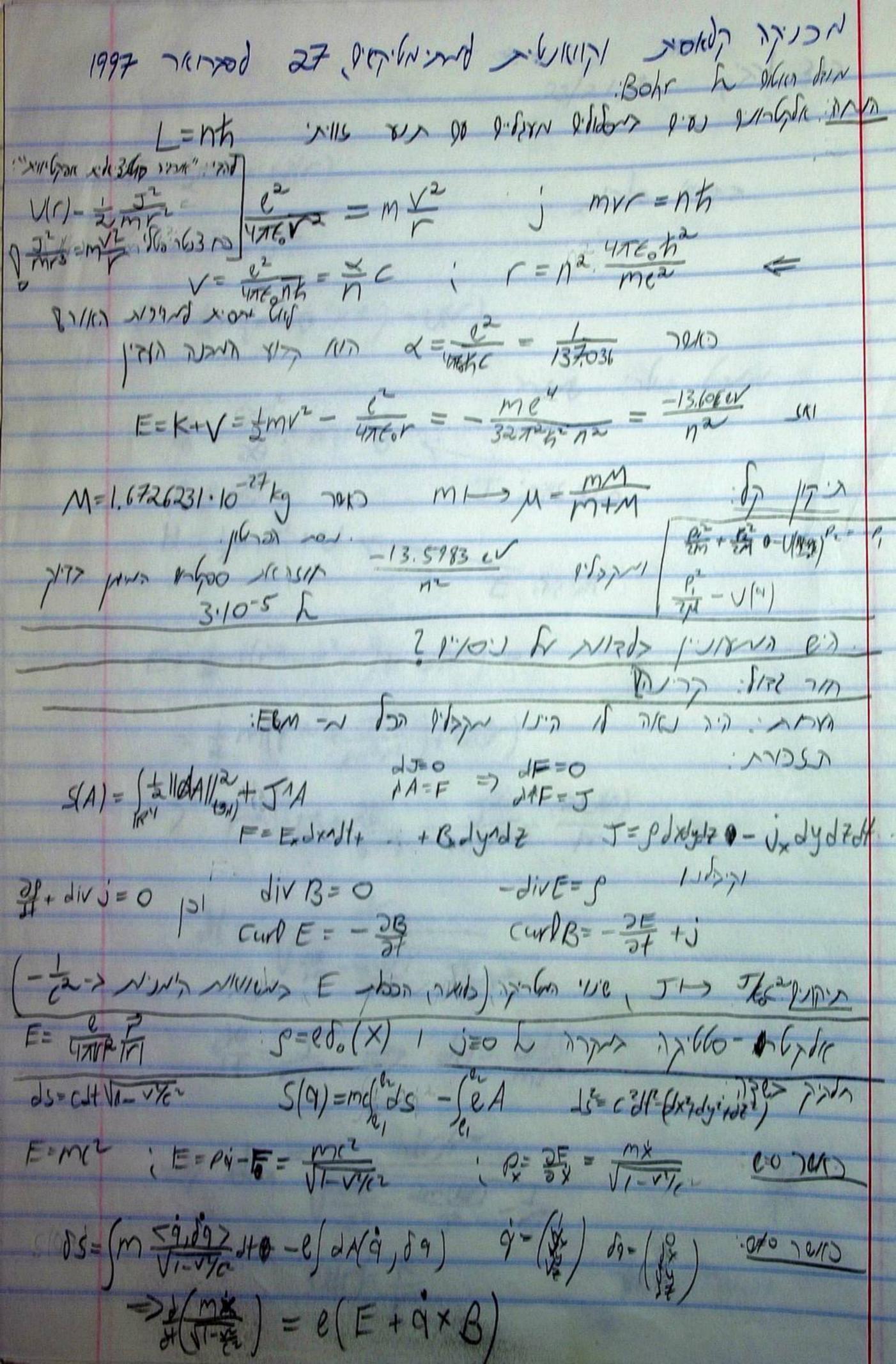
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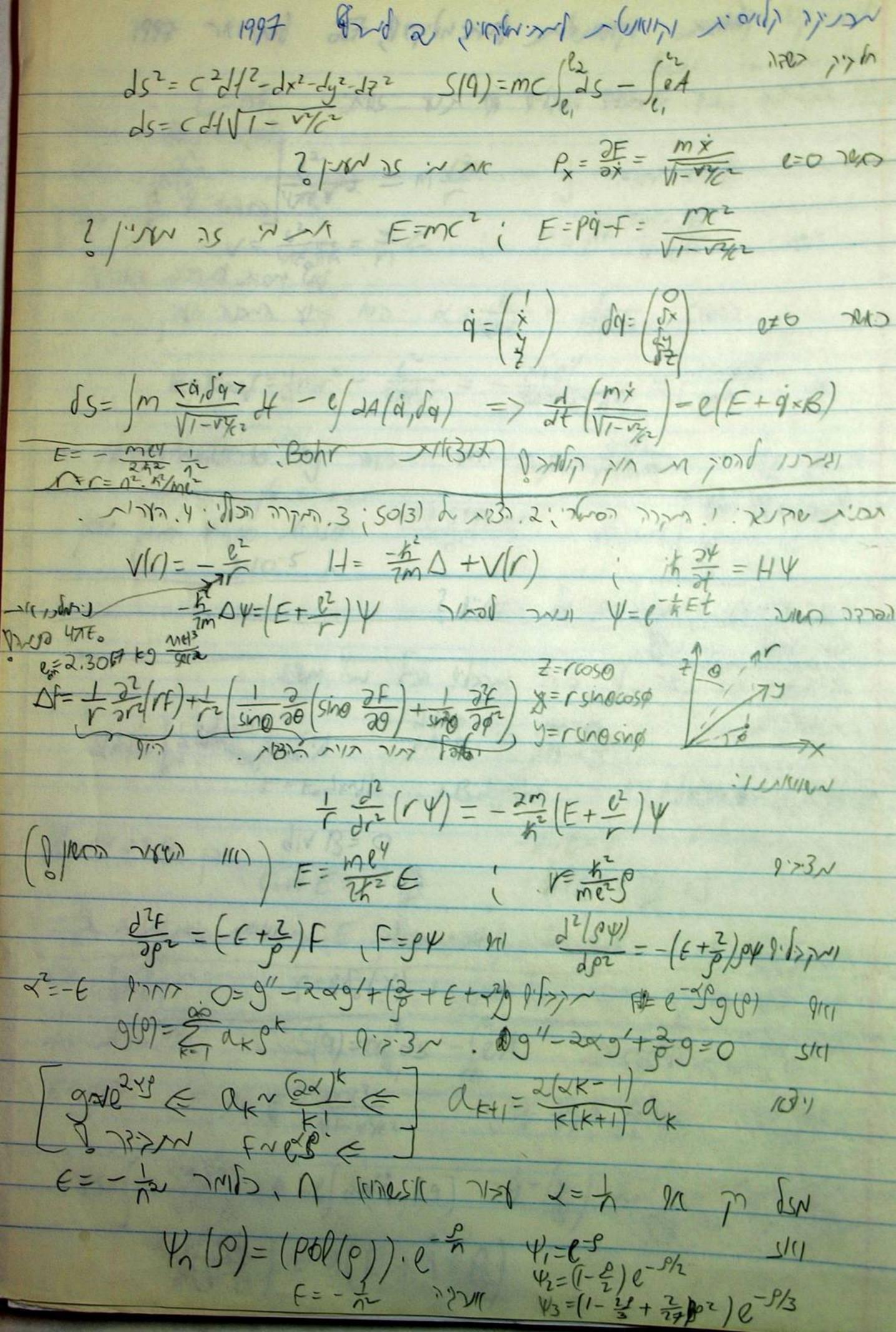
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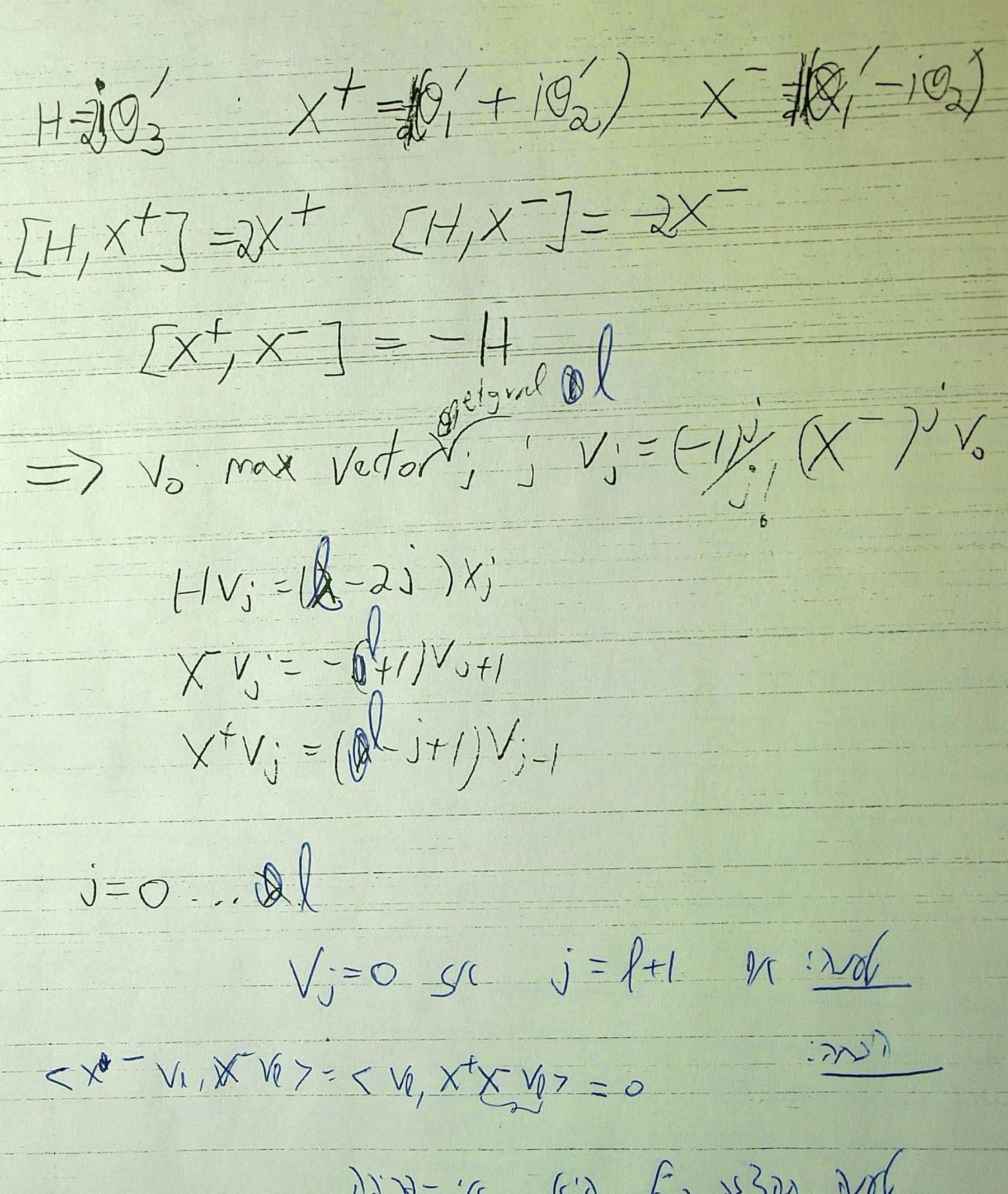
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Moth 117, April 27 1994 Just a little on Rotations & spin  $\{P_i, g_i\} = F_i \longrightarrow L^2(\mathbb{R}^3) \oplus \dots ; L^2(\mathbb{R}^3 \longrightarrow C^*)$ How does one quantite angular momenta? LOX = 94/2 - 97/9  $(\phi = 9 \times P)$ LOy= 97Px -9XPz LOZ= 9x ofy -9y/x (0x, 0y)= (9y)=-97/2, 97/2-9x/2?= 9y/x-7/y =-07 204 025 = EX (0;, P; ) = ... 202,0x4=0y 20j, P; 1 - ... Rasy quantization: LØ; HOZIPK-QKLj ( THO; ) Get [0;', 0;'] = 0/k Probi classify reps of [0i,0i]=0x in C by anty-soft-adjoint matrices. Solin: One in each dim; (+ direct sums) 1 spin-0 2 Sppn-1/2 Williams, 7 3 spin 1 cont. on ofter cido.



1997 July 6 but, ming v. Mings V. Oly J.DV (50(3)) [L'; L';]=Lx : [Noton - 611 183)  $H=2iL_3$   $X^{\pm}=L_1\pm iL_2$ [H, X+]= 2X+ (H,X)=-2X  $\left[ X^{\dagger}, X^{-} \right] = -H$ Hvo = (Vo V:= (1)) vo V HV;=(1-2j)X; X Vj=-(0+1) Vj+1 X+V=(1-j+1)Vj-1 JAN D - KER 4,-ec./e 20(3) Y 10534 : USISH 1.751412 (xfg2-xi31) ( 1554 1.751412 : 61 10 : UNRICI NERT ( 16:5-(xfg2-xi31) ( 15.54 1.751412 : 61 dim H= (1+2) -1-1(x,+ix)=((.):7xxxx He=V2+1 Ye-j=(ix)) Ye Vi'- Ye=(x,+ixz) 'i' Z=10050 PYXUM NY/ X=rsinocosp 1m(0,0)= cimp Pm (coso) y=rsinosinø 13cy (U) Pn(s)= (1-52) = 20-m (1-52) 7010 7518 de 14/62 h

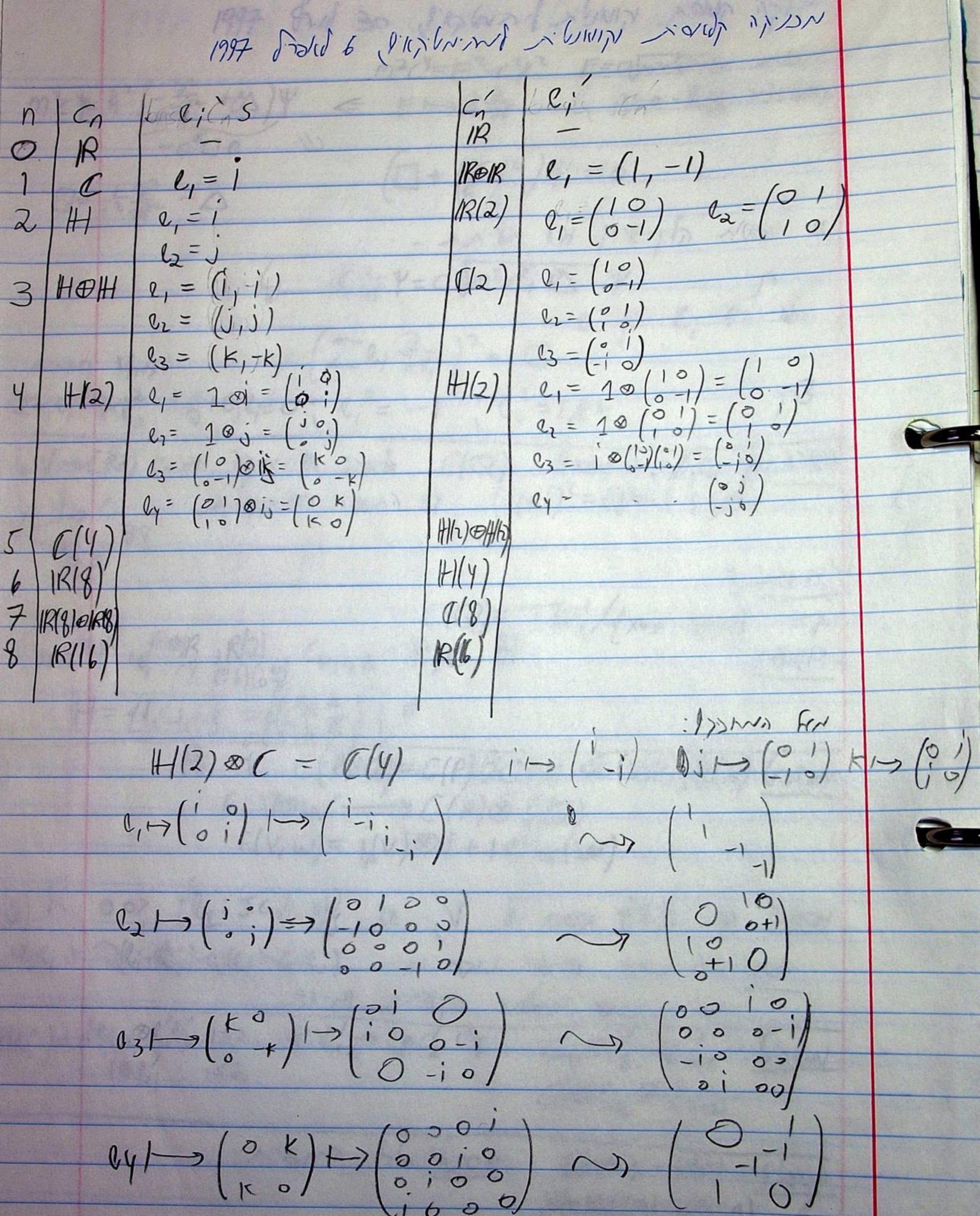
1997 Frd 13, Pigibind i Willy iont 1997 X+V;=(l-j+1)V;-1 Vo= (X,+1x)) : > 40 +(l-2j)2+=(+(j+1)·(l-j)+(l-j+1)j)= - (((+1)) - ) Lis person & miss plan del Lo poly ヤートのイクラウ、ヘンノ、ムナーナーディイクリーナン(ショング) + 1/2(rR)=-2m(E+e2-1/2/R)=-2m(2) 7=-6 or 9=0 of (F=SR)  $(=\frac{K^2}{mv})$  (1) (1pl E=-元 (本) MIND ( ) 新 1 + 15 cmm coli. 12 cmm coli.

1997 for 11 (1:17:60:m) ~ (1111) ~ (10) 1/201 YEAR ; Y=R(n); Y(0,0) TUD, R TH TELTE INVEN Tork(rR)=-2m (E+=- h(0+1))R  $3^2 = -E$   $9 = e^{2S}f$  ( F = SR )  $F = \frac{h}{2h}E$  :  $\int 3^{1}/\sqrt{e} \int \frac{de^{2s}}{2h}$ 20 =0 ak+1 = 2(k+1)-1(P+1) ak JHKN prestario 2 NIN 000 1/1 Of (BUILLING アルレン W. AV= >V MICY MY XE = X +E V, BV> May 36 33. M. May Wall J. May 2014 J. May E= VI-Vya = mc2 (1+1/2+---) -MC2 (I+(P)2 P= W/-V/N -) # 1-P/n4-P/2 = mc2(1+ = (ph)2 - = mc2 + = p2 - = p / m3c2 + DE= < Y, - 1 P W2 Y) = - 2502 < Y/E 2-2EV+V34/>= 

 $\lambda_{e} = \lambda_{o} + \epsilon \lambda_{f} + \epsilon$ Ve = Vo +EV, + \_. the 11=1 (A+FB) VE = XEVE AV = DV O= &BVo+AV, # 201, Who ) Vo) < BV., V. 7 = 1, 11V6 112

1997 ful pao 1/1/20 burn - will sold 1/1000 ind my il 1. 1/2/2 (1) 1/2/2 200 ECT DIM. -M.R. CEDIN CELLES 5612 AH50 = 2m27273 L'S : Mor-100 M37611.2 Lif Litsi (Lt) = L2+2L15+32 (L·S) Y = #((t(lt+1) - ((l+1) - s(s+1))) SU(1) > SO(3) L SO(3) L NOS) (8) (1) NOS) L NOS) L

1997 And 30 (1/9/6/201) 201 2010 17/2010 m24= 42 (-3+2+02) 4 = E1> 1/3+ (P; 1-3-1/3) (M)  $\Box = \frac{1}{2\pi} - \Delta$   $\Box + \frac{mc}{h} / \psi = 0$ 51 MOLIDY (4)-1-8/21/ 11/1 XA.V.X  $\frac{7}{2} \frac{1}{16} \frac{2}{3} + \frac{2}{12} = \sqrt{m^2 + \sqrt{2} + \sqrt{2}}$   $\sqrt{2} \frac{1}{3} \frac{2}{12} + \sqrt{2} \frac{1}{12} = \sqrt{2} \frac{1}{12} \frac{1}{12} \frac{1}{12}$   $\sqrt{2} \frac{1}{3} \frac{2}{12} + \sqrt{2} \frac{1}{12} \frac{1}{1$  $C_{0|1,2} = |R|_{1} \frac{R \otimes |R|}{R \otimes |R|_{1}} \frac{R(2)}{R(2)} C_{0|1,2} = |R|_{1} \frac{R}{R} \frac{R}{R$  $[H = d1, i, j, k] = d(-\frac{\pi}{2} \frac{\pi}{2})$   $(C(P \oplus Q) = C(P) \hat{Q} C(Q) \in (V, P); (V, Q) Gen$ F: VOW -> C(P) & CQ) F(V, W) = 1(V) 01+101@(W) C(Q) h 000 Ply: Icti. 11/2 SI (V & 0:00 2 2 2:3 91 Drow (is)+isi; = Q(li+6i)-Q(li)-Q(li) -~ you inor , 7157: n=1:70/1 1-4-17 NGS/2 WILL 14-FU Y(e;)= de; 20e; e'x isa Y: Vn+2 > (06; (1) Cn06= Cn+2 (sen) COOC = CA+2 1. 1/2 /2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1/2 | 1R(n) 8/R(m) = 1R(mn) COC = COC H-00= c(2) HOH = 1R(4)



1997 Field 10 Dischient prosent in post 1970 . 7/2 1/31) 1 : Dirac 1/1/2/2 ik py=mczy ih (== ==) y= mczy 86 4=mczy Zcip; 4 = mczy  $f_i \rightarrow f_i - eA_i$ : ·fin 3.3 (1) (1) - (V) / (1)

-1997 Ford 13 styllwish ibusin 2010 17000 AV=mc V :200 1600 17000

大Y=mc y : 100 在=-ih3=-ch 70'3 Nater : (:= (0 0i-) lo= = -(0 1) (-ih2+-eA) Y+= Qo-(ih2x;+eA;) Y-+mc2Y+ (ihof+cAo) Y== C Zoi+(ikox;+cA) K+ +NCZY (-ikay-eAo) \( \varphi = C \( \sigma\_i \) (ikaxite Ai) \( \varphi = \varphi \) (ihattlA.) \( = C\( \int\_{\text{2}} \) (ikax;+eA;) \( \frac{1}{4} + 2mc^2 \)

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Marili Gerox Milling presention, 1 cm, 2661. (-ikaz-eAo) \( \frac{1}{4} = \frac{1}{2} \sigma\_{i,j} \sigma\_{i-1} \sigma\_{i+1} (ikaz; +eAi)(ikaz; +eAi)(ikaz; +eAi) \( \frac{1}{4} \) TI- TI+ + TI- TI+ = 0 (TITH = -1 1001  $Y_i y_i = Y_k$   $\int_{Y_i} y_i = Y_k = \sigma_i - \sigma_{i+1}$ 1 = -1 1511 = 0 = 0 = 0 = 0 = - 1 = - 1k = \frac{1}{m} (-\frac{1}{(ikdx,+(A;)^2)}+\frac{1}{2}\&k[ikdx,+(A;)]\}

ment they kind in presiding the puri Ty = mcy T<sub>x</sub> = -ih= (-ehx): 12 pt 11102 (loPo+ Zeili) Y=mcY EZPY = MCRY-DE COCIPY = MOST MCCSY-ZFIPIY lof; = 1 te; f; lo = -te; => dlo, f; } = 0  $e^{is}(mce_{o} + \Sigma p_{i}f_{i})e^{is} = (0) = \lambda e_{o}$ 

1.1 Formulation of a Relativistic Quantum Theory

Since the principles of special relativity are generally accepted at this time, a correct quantum theory should satisfy the requirement of relativity: laws of motion valid in one inertial system must be true in all inertial systems. Stated mathematically, relativistic quantum theory must be formulated in a Lorentz covariant form.

In making the transition from nonrelativistic to relativistic quantum mechanics, we shall endeavor to retain the principles underlying the nonrelativistic theory. We review them briefly:

1. For a given physical system there exists a state function  $\Phi$  that summarizes all that we can know about the system. In our initial development of the relativistic one-particle theory, we usually deal directly with a coordinate realization of the state function, the wave function  $\psi(q_i \cdots, s_i \cdots, t)$ .  $\psi(q,s,t)$  is a complex function of all the classical degrees of freedom,  $q_1 \cdots q_n$ , of the time t and of any additional degrees of freedom, such as spin  $s_i$ , which are intrinsically quantum-mechanical. The wave function has no direct physical interpretation; however,  $|\psi(q_1 \cdots q_n, s_1 \cdots s_n, t)|^2 \geq 0$  is interpreted as the probability of the system having values  $(q_1 \cdots s_n)$  at time t. Evidently this probability interpretation requires that the sum of positive contributions  $|\psi|^2$  for all values of  $q_1 \cdots s_n$  at time t be finite for all physically acceptable wave functions  $\psi$ .

2. Every physical observable is represented by a linear hermitian operator. In particular, for the canonical momentum  $p_i$  the operator correspondence in a coordinate realization is

$$p_i \rightarrow \frac{\hbar}{i} \frac{\partial}{\partial q_i}$$

3. A physical system is in an eigenstate of the operator  $\Omega$  if

$$\Omega \Phi_n = \omega_n \Phi_n \tag{1.1}$$

where  $\Phi_n$  is the *n*th eigenstate corresponding to the eigenvalue  $\omega_n$ . For a hermitian operator,  $\omega_n$  is real. In a coordinate realization the equation corresponding to (1.1) is

$$\Omega(q,s,t)\psi_n(q,s,t) = \omega_n\psi_n(q,s,t)$$

J. Springer, Berlin, 1933. L. I. Schiff, "Quantum Mechanics," 2d ed., McGraw-Hill Book Company, Inc., New York, 1955. P. A. M. Dirac, "The Principles of Quantum Mechanics," 4th ed., Oxford University Press, London, 1958.

4. The expansion postulate states that an arbitrary wave function, or state function, for a physical system can be expanded in a complete orthonormal set of eigenfunctions  $\psi_n$  of a complete set of commuting operators  $(\Omega_n)$ . We write, then,

$$\psi = \sum_{n} a_{n} \psi_{n}$$

where the statement of orthonormality is

$$\sum_{n} \int (dq_1 \cdot \cdot \cdot) \psi_n^*(q_1 \cdot \cdot \cdot \cdot, s \cdot \cdot \cdot \cdot, t) \psi_m(q_1 \cdot \cdot \cdot \cdot, s \cdot \cdot \cdot \cdot, t) = \delta_{nm}$$

 $|a_n|^2$  records the probability that the system is in the nth eigenstate.

5. The result of a measurement of a physical observable is any one of its eigenvalues. In particular, for a physical system described by the wave function  $\psi = \sum a_n \psi_n$ , with  $\Omega \psi_n = \omega_n \psi_n$ , measurement of a physical observable  $\Omega$  results in the eigenvalue  $\omega_n$  with a probability  $|a_n|^2$ . The average of many measurements of the observable  $\Omega$  on identically prepared systems is given by

$$\langle \Omega \rangle_{\psi} \equiv \sum_{s} \int \psi^{s}(q_{1} \cdot \cdot \cdot \cdot , s \cdot \cdot \cdot \cdot , t) \Omega \psi(q_{1} \cdot \cdot \cdot \cdot , s \cdot \cdot \cdot \cdot , t) (dq_{1} \cdot \cdot \cdot \cdot )$$

$$= \sum_{n} |a_{n}|^{2} \omega_{n}$$

6. The time development of a physical system is expressed by the Schrödinger equation

$$i\hbar \, \frac{\partial \psi}{\partial t} = H\psi \tag{1.2}$$

where the hamiltonian H is a linear hermitian operator. It has no explicit time dependence for a closed physical system, that is,  $\partial H/\partial t = 0$ , in which case its eigenvalues are the possible stationary states of the system. A superposition principle follows from the linearity of H and a statement of conservation of probability from the hermitian property of H:

$$\frac{d}{dt} \sum_{s} \int \psi^* \psi(dq_1 \cdot \cdot \cdot) = \frac{i}{\hbar} \sum_{s} \int (dq_1 \cdot \cdot \cdot) [(H\psi)^* \psi - \psi^* (H\psi)]$$

$$= 0 \qquad (1.3)$$

We strive to maintain these familiar six principles as underpinnings of a relativistic quantum theory.